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Balancing Equations. 1. Do not EVER change the formulas of any of the chemical compounds! 2. If it doesn't work, start over from scratch. 3. Always do the inventory for every single step! 4. One step at a time. 5. If the equation simply won't balance, just write a "2" in front of the most ...

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Equations Worksheet - Answers Note to students: It is acceptable to leave spaces blank when balancing equations - blank spaces are interpreted as containing the number "1".

1)  $1 \text{ Na}_3 \text{ PO}_4 + 3 \text{ KOH} \rightarrow 3 \text{ NaOH} + 1 \text{ K}_3 \text{ PO}_4$

2)  $1 \text{ MgF}_2 + 1 \text{ Li}_2 \text{ CO}_3 \rightarrow 1 \text{ MgCO}_3$

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## **Answers | Mychaume.com**

balancing equations. Doing the balancing equations race: Divide the students into teams of two. (For honors classes, have each student work on his or her own.) The kids should be sitting side-by-side so they can work together more efficiently. Pass out one worksheet face down for every student. At the count of three, have the students start ...

## **Balancing Equations Race**

A balanced chemical equation gives the number and type of atoms participating in a reaction, the reactants, products, and direction of the reaction. Balancing an unbalanced equation is mostly a matter of making certain mass and charge are balanced on the reactants and products side of the reaction arrow.

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